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Chapter 14 Acids
And Bases

Chapter 14 Acids And Bases

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Chapter 14 Acids And Bases

an acid that is composed of H, O, and a third element. The H & O are covalently bonded and are called the hydroxyl group.

Property of bases.

taste bitter, feel

slippery, change color

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Chapter 14 Acids And Bases

of indicators, react with acids to produce a salt and water
neutralization rxn, are electrolytes. Arrhenius or traditional base.

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(14.1) The nature of acids and bases (14.2)

Acid strength (14.3)

The pH scale (14.4)

Calculating the pH of strong acid solutions

(14.5) Calculating the

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pH of weak acid
solutions (14.6) Bases
(14.7) Polyprotic acids

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Chapter 14 - Acids and Bases . 14.1 The Nature of Acids and Bases . A. Arrhenius Model 1. Acids produce hydrogen ions in aqueous solutions 2. Bases produce hydroxide ions in aqueous solutions B. Bronsted-Lowry Model 1. Acids are proton donors 2. Bases are proton acceptors 3. H₃O⁺ is called the hydronium ion C. Conjugate Acid- Base

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Chapter 14 - Acids and
Bases. Description.

Chapter 14 Review.

Total Cards. 36.

Subject. Chemistry.

Level. 11th Grade. ...

General Properties of
Acids: Definition. 1)

sour taste, corrosive 2)

characteristically

changes acid-base

indicator 3) react with

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active metals to form hydrogen gas 4) react with bases to form salt and water 5) conduct ...

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ANS: e) ClO_4^- , F^- ,
 OAc^- , CN^- PAGE: 14.2
11. The conjugate base
of a weak acid is a) a
strong base b) a weak
base c) a strong acid d)
a weak acid e) none of
these ANS: b) a weak
base PAGE: 14.2 12.
Which of the following
is the equilibrium
constant expression for
the dissociation of the
weak acid HOCl ? a) $K =$

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CHAPTER 14 Acids and Bases

Modern Chemistry
Chapter 14 Vocab
Acids and Bases.

binary acid. oxyacid.
Arrhenius acid.

Arrhenius base. an acid
that contains only two
different elements:

hydrogen an.... an acid
that is a compound of
hydrogen, oxygen, and
a third el.... a chemical
compound that

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increases the
concentration of
hydrog....

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CHAPTER 14 REVIEW

Acids and Bases

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

Name the following compounds as acids:

sulfuric acid a. H_2SO_4

sulfurous acid b. H_2SO_3

3 hydrosulfuric acid c.

H_2S perchloric acid d.

HClO_4 hydrocyanic

acid e. hydrogen

cyanide 2. $\text{H}_2\text{C}_2\text{O}_4$

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14 Acids and Bases - Password

acid, and water acts as the base. This is the acid ionization

reaction. $HA(aq) + H_2O(l) \rightleftharpoons A^-(aq) + H_3O^+(aq)$

6) When a base is added to water, there is a Bronsted-Lowry acid base reaction where the base behave as a base and water acts as the acid. This is the base ionization reaction. Be

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able to identify or provide the conjugate acids and ...

Chapter 14: Acids and Bases

CHAPTER 14 ACIDS AND BASES 501 24. A Lewis acid must have an empty orbital to accept an electron pair, and a Lewis base must have an unshared pair of electrons. 25. $\text{NH}_3 + \text{NH}_3 \rightleftharpoons \text{NH}_2^- + \text{NH}_4^+$ Acid Base Conjugate Conjugate

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Base Acid

CHAPTER 14 ACIDS AND BASES - HCC Learning Web

Major topics: Arrhenius vs. Bronsted-Lowry definition of acids and bases, conjugate acid/base, acid dissociation constant (K_a), & strong vs weak acids

Chapter 14 (Acids and Bases) - Part 1 - YouTube

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Chapter Fourteen:
Acids and Bases
Chapter interactives:
Acids/bases and
indicators Chemistry
battleship Chemistry
jokes pH simulation
Chapter Homework:
Section 1: Chapter
review 1 thru 7.
Section 2: Chapter
review 12, 13. Section
3: Chapter review 19
thru 25.

Chapter Fourteen [Acids and Bases]

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Chapter 14: Acids and

Bases 1. Acids and

Bases. 1. Which of the
following is the

equilibrium constant
expression for the

dissociation of the
weak acid HOCl?

e) none of these. 2. In

which of the following
reactions does the

H_2PO_4 ion act as an

acid? d) The ion cannot
act as an acid.

**Chapter 14: Acids
and Bases 1 - Docest**

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Chapter 14 - Acids and Bases Pablo Gonzalez. Loading... Unsubscribe from Pablo Gonzalez? ... Chapter 16 Acid-Base Equilibria - Duration: 1:06:16. Michael Farabaugh 22,088 views.

Chapter 14 - Acids and Bases

Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In this chapter you will learn The definitions

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and identity of substances that are called acids Chapter-14 -Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In...

Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In ...

Acids and Bases:
Chapter 14 & 15 . HW:
•Read Ch 14: •Fill in as much of the acid base table as you can, as you read , Acid base

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conductivity and
reactivity Conductivity,
Reac'tivity,,
Hydrochloric*acid* high
high Acidic*acid* **
low* medium
Dis4lled*water* none*
none*
Ammoniumhydroxide
med* none*
Sodiumhydroxide* high
none* *

Acids and Bases: Chapter 14 & 15

Friday, January 19 -

Day A Agenda: 1) Test

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- Chapter 12 and 13
HW: 1) Watch Videos
and take notes on intro
to acids and bases ,
and pH scale 2) Read
14.1- 14.4

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