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It states that thermal energy can flow from colder objects to hotter objects only if work is done on the system. Third law of thermodynamics It states that

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absolute zero cannot be reached.

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16.1 Thermal Energy and Matter Heat flows spontaneously from hot objects to cold objects. • Heat is the transfer of thermal energy from one object to another because of a temperature

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difference. Temperature is related to the average kinetic energy of the particles in an object due to their random motions through space.

Chapter 16 Thermal Energy and Heat

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thermal energy from one object to another as the result of a difference in _____ . _____ produces heat.

Physical Science Chapter 16: Thermal Energy and Heat ...

Chapter 16 Thermal Energy and Heat
Section 16.1 Thermal Energy and Matter

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(pages 474-478) This section defines heat and describes how work, temperature, and thermal energy are related to heat. Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained.

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or cold an object is compared to a reference.... Metric temperature scale on which water freezes at 0 degrees a....

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