

## Che 512 Course Notes Chapter 2 2 Characterization Of

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### Che 512 Course Notes Chapter

CHE 512 Course Notes Chapter 2 (Updated 01/05) 6 2.1.8 Dimensionless Representation It is customary to use dimensional time  $\theta = t/\tau$ . Then the set of function defined over the  $\theta$  domain are  $E_0, F_0, I_0, A_0$ . Since  $(\theta)$  Fraction of the fluid of residence time  $t$  to  $t + dt = (\theta)$  Fraction of the fluid of residence time  $\theta$  to  $\theta + d\theta$   
 $E(\theta)dt = E_0(\theta)d\theta$

### Che 512 Course Notes Chapter 2 2. CHARACTERIZATION OF ...

CHE 512 – Chapter 7 Spring 2005 1 2. Basic Notions of Mixing and Turbulence By mixing we understand the process of achieving a uniform composition by forced and random movement of fluid elements. Chemical reactions cause compositional differences in process vessels. Gross fluid motion reduces the scale over which such differences exist.

### CHE 512 - Chapter 7 Spring 2005

CHE 512 – Chapter 12 Spring 2005 1 Treatment of Very Rapid Reactions - Summary of Basic Concepts We have so far considered only the situations where the characteristic reaction time is much larger than the mixing time,  $R_{mix}$ . Now let us look at the other extreme

### CHE 512 Chapter 12 - classes.engineering.wustl.edu

CHE 512 – Chapter 9 Spring 2005 1 Modeling Micromixing Effects in a CSTR CSTR, of all well behaved reactors, has the widest RTD i.e  $\sigma^2 = 1$ . This means that large differences in performance can exist between segregated flow and operations at maximum mixedness conditions. The easiest thing to treat is the case of a premixed feed.

### CHE 512 - Chapter 9 Spring 2005

CHE 512 – Chapter 8 Spring 2005 3 1  $V \frac{\partial \psi}{\partial t} + \frac{\partial}{\partial i} = 1 \frac{\partial \zeta_i}{\partial t} m \sum (\psi V_i) + D - B = 1 V (Q) \text{in} \psi - Q \text{out} \psi$  (2) where  $V_i = d\zeta_i / dt$  At steady state  $\frac{\partial}{\partial i} = 1 \frac{\partial \zeta_i}{\partial t} m \sum (\psi V_i) + D - B = 1 V (Q) \text{in} \psi - Q \text{out} \psi$  (3) Examples: 1. Take  $\zeta_1 = \alpha$  – age of fluid element  $m = 1$  steady state  $Q_{in} = Q_{out} d \alpha \psi d \alpha dt + D - B = 1 V$

### CHE 512 - Chapter 8 Spring 2005

CHE 512 – Chapter 7A Spring 2005 7 Let us consider now the solution of eq (24a). Since the lamella is constantly shrinking in the  $x'$  direction, i.e., its thickness  $\delta$  is being reduced in time, it is useful to immobilize this moving boundary by introducing new dimensionless coordinate  $x' z \delta = (25)$

### Lamellar Model for Mixing and Reaction

CHE 512 – Chapter 6 Spring 2005 6 There is a forced vortex zone ( $J_{ur} = \omega$  in a range of 70% of  $d_l$  and a quasi-free vortex zone  $2 u_{rc} r \omega =$  in the outer part, for case of water. With increase in  $\mu$ ,  $r_c$  is reduced and becomes zero at the transitional viscosity from turbulent to laminar case. The transition is a function of the power

### I. Stirred Tank Reactor

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CPB 412/512 Dr. Paluch Chapter 5 Notes If we think back to CPB/MME 314 Engineering Thermodynamics, Chapter 5 will pertain to a single component in a homogeneous closed system. (We can and will extend the material to heterogeneous open systems and to mixtures in later chapters.) The Gibbs Phase Rule tells us that for a single component, single phase system, two intensive thermodynamic ...

### chapter 5\_notes.pdf - CPB 412/512 Dr Paluch Chapter 5 ...

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CPB 412/512 Dr. Paluch Chapter 5 Notes 2 Residual Properties We define a residual property as an actual property relative to an ideal gas at the same conditions. Let  $F$  be an arbitrary state function with independent variables  $x$  and  $y$ , then  $F^R(x,y) = F(x,y) - F^{ig}(x,y)$  (1) where the superscript "R" and "ig" correspond to the residual and ideal gas properties, respectively.

### chapter 5\_notes\_2.pdf - CPB 412/512 Dr Paluch Chapter 5 ...

8/26/2020 Ch 01 HW 9/12 Correct Chapter 1 Check Your Understanding Question 6 Part A What is the role of the serous membranes covering some organs? ANSWER: Correct The two serous membrane layers are separated by a cavity containing serous fluid. This fluid allows organs to slide without friction across the cavity walls and one another as they carry out their routine functions.